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Patent application number (The Patent Office will fill in this part)

13 NOV 2002

0226448.9

Full name, address and postcode of the or of each applicant (underline all surnames)

CLARIANT INTERNATIONAL LTD Rothausstrasse 61

CH-4132 Muttenz

06971634001 Patents ADP number (if you know it)

If the applicant is a corporate body, give the country/state of its incorporation

Switzerland

Title of the invention

Organic Compounds

Name of your agent (if you have one)

"Address for scrvice" in the United Kingdom to which all correspondence should be sent (including the postcode)

CLARIANT UK LTD Attn. Ms. Anne Plowden Calverley Lane Horsforth/Leeds, LS18 4RP

Patents ADP number (if you know it)

7156086001

07156086003

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Country

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Yes

- any opplicant named in part 3 is not an inventor, or
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Description

Claim(s)

Abstract

Drawing (s)

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Priority documents

Translations of priority documents

Statement of inventorship and right to grant of a patent (Patents Form 7/77)

Request for preliminary examination and search (Patents Form 9/77)

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11.

I/We request the grant of a patent on the basis of this application.

Signature

Clariant International Ltd Date

November 13, 2002

Dr. Dieter Dünnwald

Patent Attorney

C. Hewsler Christel Housler

Patent Administrator

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PATENTS FORM 1/77

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ORGANIC COMPOUNDS

The invention relates to novel dyestuff and mixtures thereof, the use of such compounds and/or mixtures for printing recording materials, especially paper or papery substrates, textile fibre materials, plastic films and plastic transparencies by the inkjet printing process and also to the recording materials printed thereby.

Inkjet printing processes are becoming more and more important for industrial applications.

Inkjet printing processes are known. In what follows, the principle of inkjet printing will only be discussed very briefly. Details of this technology are described for example in the Ink-Jet-Printing section of R.W. Kenyon in "Chemistry and Technology of Printing and Imaging Systems", Peter Gregory (editor), Blackie Academic & Professional, Chapmann & Hall 1996, pages 113-138, and references cited therein.

In the inkjet printing process, individual droplets of the ink are sprayed from a nozzle onto a substrate in a controlled manner. The continuous inkjet method and the drop-on-demand method are employed predominantly for this purpose. In the case of the continuous inkjet method, the droplets are produced continuously and droplets not needed for printing are diverted into a collecting vessel and recycled. In the case of the discontinuous drop-on-demand method, by contrast, droplets are generated and printed as desired, i.e. droplets are only generated when this is necessary for printing. The droplets may be generated for example by means of a piezo inkjet head or by means of thermal energy (bubble jet).

By additionally disposing at least one nozzle with yellow, magenta or cyan ink side by side it is possible to obtain colour reproductions in high quality. This process is known as polychromatic printing or, when three colour components are used, as trichromatic printing.

The composition of the invention can be used with all known and suitable inkjet printers for printing paper or papery substrates, textile fibre materials, plastic films and plastic

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The composition of the ink for the inkjet printing process has to possess a suitable conductivity, sterility in storage, viscosity and surface tension to meet the specific requirements of inkjet ink. In addition, the prints on the recording materials have to have good properties and fastness.

transparencies. This applies not only to the use in monochromatic printing but also to

polychromatic printing, especially trichromatic printing.

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Useful recording materials, as mentioned above, are preferably paper and papery substrates, textile fibre materials, plastic films and plastic transparencies. But glass and metal may be used as well.

- Useful papers or papery substrates include all known such materials. Preference is given to papers or papery substrates coated on at least one side with a material which is particularly receptive to ink compositions. Such papers or papery materials are described inter alia in DE 3018342, DE 4446551, EP 164196 and EP 875393.
- Useful textile fibre materials are in particular hydroxyl-containing fibre materials. Preference is given to cellulosic fibre materials, which consist of or comprise cellulose. Examples are natural fibre materials such as cotton, linen or hemp and regenerated fibre materials such as, for example, viscose and also lyocell.
- Useful plastic films or plastic transparencies include all known such materials. Preference is given to plastic films or plastic transparencies coated on at least one side with a material which is particularly receptive to the ink compositions. Such plastic films or plastic transparencies are described inter alia in EP 755332, US 4935307, US 4956230, US 5134198 and US 5219928.

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This invention provides dyestuff according to the formula (I)

 R_3 R_4 R_5 R_5 R_6 R_7 R_8 R_7

wherein

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R₁, R₂, R₃, R₄, independently from each other signifies H, SO₃H, SO₂R, SO₂NRR", COOR, COOH, OH, alkoxy, NCOCH₃, or NR'R", or two of the substituents R₁, R₂, R₃, or R₄ from together a ring which is annealed to the benzthiazole moitie

R₅, R₆, R₇, R₈, independently from each other signifies H, Methoxy, Alkyl, Aryl, Alkoxy, SO₃H, SO₃R, Cl. Br, F, NR'R", SO₂-CH₂-CH₂-OH, or or two of the substituents R₅, R₆, R₇, or R₈, form together a ring which is annealed to the phenyl group of the aminopyrazolo moitie.

R, R' and R'' are independently from each other H, alkyl or alkoxy or arylgroups groups.

Alkyl and alkoxy groups referred to above are preferably radicals

Alkyl and alkoxy groups referred to above are preferably radicals with up to 8 carbon atoms, more preferably up to 4 carbon atoms, e.g., butyl and more preferably 1 or 2 carbon atoms; any substituents thereon may be halogen, hydroxy, cyano, alkoxy, acyl, acyloxy or acylamino. Those alkyl or alkoxy groups having 3 or more carbon atoms may be straight-chain, branched or cyclic, for example, cyclohexyl.

The invention provides in another of its aspects a process for the production of the dyestuffs according to the formula I wherein a diazotised amine of formula II

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is coupled with a compound of formula III:

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The compounds of formulae II and III are known starting materials, although they may also be readily synthesised from commonly available reagents via known syntheses. Diazotisation and coupling are effected in the usual manner. Preferably the amine II is diazotised at a temperature of from 0 DEG to 20 DEG C. and at a pH of from 0 to 2, more preferably 0 to 1. The coupling reaction is preferably carried out at a temperature of from 0 DEG to 40 DEG C. and at a pH of from 0 to 7, more preferably 0 to 4.

Dyestuffs according to the formula (I) may be isolated from the reaction medium according to known methods, for example, by salting out with an alkali metal salt, filtering and drying optionally in vacuo at a slightly elevated temperature.

Depending upon the reaction and/or isolation conditions, the dyestuffs according to the formula (I) may be obtained as the free-acid, salt or mixed salt form, containing, for example, one or more cations selected from the alkali metal cations, e.g., the sodium ion, or an ammonium ion or an alkylammonium cation, e.g., mono-, di-, or tri-methyl or ethylammonium cations. The dyestuffs may be converted from the free-acid form to a salt or mixed salt form or vice-versa or from one salt form to another according to conventional techniques.

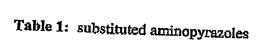
The compounds of formula (I) may be converted into dyeing preparations. Processing into stable liquid, preferably aqueous, or solid (granulated or powder form) dyeing preparations may take place in a generally known manner.

A further embodiment of the invention relates to the use of the above mentioned compositions for the ink-jet printing process.

10 In the following examples, the temperatures are given in degrees celsius. The percentages and parts are understood to be by weight.

EXAMPLE

173g of methanilic acid were diazotazed at 0 to 5 °C and reduced by 269g Sodium sulfite in the presence of 40 sodium hydrogenearbonate at 14 to 16°C, after this step the product was hydrolized by adding 330g sulfuric acid (85%) at 75°C. The resulting hydrazine was filtered off and at pH 7 reacted with 82g 3-Aminocrotonic nitrile. The product was presipitated by adding 350g hydrochloric acid, filtered off, washed and dried which yields a press cake of 270g of 1-(3'-sulfophenyl)-3-methyl-5-aminopyrazol.



Code	R ₅ '	R ₆ *	R7'	Rg	R9'
1	SO ₃ H	-	-	-	
2	CH ₃		SO₃H	_	
3	SO ₃ H	-		SO ₃ H	
4	89	SO₃H	-	-	
5	-	SO₃H	NH ₂	_	
6	-	SO₃H	OCH ₃	-	
7	-	CH₃	SO ₃ H	-	_
8	-	NH ₂	SO ₃ H	-	n.
9	-	OH OH	-	-	-
10		_	SO ₃ H	-	_
11	-	-	СООН	-	
12	-	-	О ОН	-	
13)	SO₃H	_	

14	SO₃H	. Ç	-	_
15	-	SO ₈ H	-	-
16	-	SO ₃ H	-	-
17	-	· So ₃ H	-	-
18	-	Hos	-	-
19	SO₃H	но₅ѕ	-	-
20	-	но, я	-	-

EXAMPLE

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218g 2-amino-nitrobenzene-5-sulfonic acid were added to 1100ml of a 1.0 Molar Sulforic acid at 5°C and diazotazed. The resulting suspension was slowly added a solution of 180ml water, 152g NH₄SCN and 5g Cu(SCN)₂ at 0 to 5 °C ant stirred for additional 3 hours at rommtemperature. The reaction mixture was poured into a hot (90 to 95°C) suspension of 1000ml water, 500g iron powder and 17g of a 37% hydrochloric acid and the resulting suspension was stirred till the reduction was completed. The mixtures was brought to a pH of more than 9 and the iron slurry was filtered off. The product was precipitated by lowering the pH, filtered off, washed and dryed, this yields 200g product (2-Amino-benzthiazole-5-sulfonic acid) as an press cake.

Table 2: substituted benzthiazoles:

Code	R ₁ '	R ₂	R ₃ *	R ₄
1	SO₃H	_	-	-
2	SO3H	-	CH₃	-
3	SO₃H	-	OCH ₃	-
4	СООН	-	-	
5	OCH₃	-	-	-
6	-	OCH₃	_	-
7	-	SO₃H	-	-
8	-	SO₃H	OCH ₃	_
9	n	SO ₂ NH ₂	-	-
10	-	-	ŚO₃H	-
11	-	**	СООН	-
12		-	COOCH ₃	•
13	-	-	COOEt	-

14	_	-	NH ₂	-
15	-	_	*NH	1
16	- -	· <u>-</u>	OCH₃	SO₃H
17	-	-	*0 OH	
18		-	*SO ₂ OH	LI .
19	-	-	*SO ₂ -N	<u>-</u>
20	-	-	-	COOH
21	-		-	SO₃H
22		-	HO ₃ S **	

5 EXAMPLE

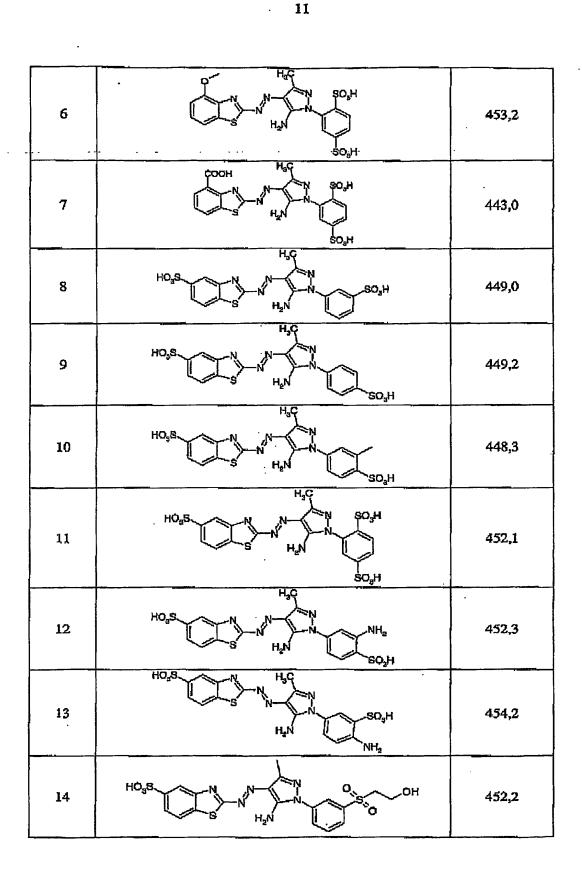
To 500g sulfuric acid (98%) 40g sodium nitrite were added and cooled to 5°C. To the cooled solution 115g 2-Amino-benzthiazole-5-sulfonic acid were added while adding 625g of crushed ice to keep the temerature low. After the diazotation of the 2-Amino-benzthiazole-5-sulfonic acid has been completed, this suspension was added to a solution of 1000ml water and 127g 1-(3'-sulfophenyl)-3-methyl-5-aminopyrazol. The pH was kept at about pH 5 to 6 by adding about 1330g of a 30% NaOH solution. The mixture was stirred until the coupling was copleted. The resulting dye was purified by diafiltration

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The affinity of this dyestuff was excellent (>95%). The migration of the dyestuff, which is decisive for the levelness of the dyeings, and its fastness, especially light and wet fastness, and also its boiling resistance, were very good.

dye Nr	Struktur	λ _{max}
1	SOJH H ₃ C N SO ₃ H	408,3
2	SO ₃ H H ₂ U SO ₃ H	404,9
· 3	SO ₃ H H ₂ C N SO ₄ H SO ₄ H	427,4
4	SO ₃ H H ₃ C N H ₂ N SO ₃ H	435,1
5	SO ² H H ² N COOH	439,7





		·····
15	HO ₃ S N SO ₉ H	462,1
16	HO _S S T N N N N SO ₃ H	462,4
17	HO'S H'N SO'H	461,7
18	HO ₀ S N N N N SO ₀ H	463,2
19	HO ₈ S N N N N N N N N N N N N N N N N N N N	460,3
20	HO _B S H ₂ N H ₂ N HO ₃ S	461,3
21	HO ₂ S H ₂ N SO ₂ H H ₀ ,S	465,8
22	H ₃ C H ₂ N H ₂ N H ₂ N H ₃ S SO ₃ H	468,2



23	HO ² S A H ² N A OH	443,7
24	H _I N SO ₃ H SO ₃ H	454,3
25	H ₂ C N SO ₃ H SO ₈ H	459,5
26	HO ₈ S SO ₃ H	453,0
27	Ho ₃ s H _e N So ₉ H	452,5
28	HO ₃ S H ₂ N N SO ₃ H	451,8
29	H ₃ C N SO ₃ H H ₂ N SO ₃ H	455,9
30	(CH ₃) ₂ -N-O ₃ S SO ₃ H	459,7
31	HOOO SO ₉ H	454,7

32	H ₃ COOC SO ₃ H S	453,2	
33	EKOOC SO ₃ H	453,8	
34	H ₂ O H ₂ N SO ₂ H	472,8	
35	H ₃ C H ₃ C N ₂₀₃ H ₂₀₃ H ₂₀₃ H	459,0	
36	HO	458,1	
37	H ₂ C N SO ₃ H H ₂ N SO ₃ H	457,4	-
38	HO'S H'N SO'H	389,7	
 39	H ₃ C N N H ₂ N SO ₃ H	371,8	
 40	H ₃ C N N N N N N N N SO ₃ H	458,1	

41	H ₃ C N S H ₂ N SO ₃ H	442,9
42	Ho ₃ s N N SO ₃ H	470,3

Claims

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1. dyestuff according to the formula (I)

 $\begin{array}{c|c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$

wherein

R₁, R₂, R₃, R₄, independently from each other signifies H, SO₃H, SO₂R, SO₂NR'R'', COOR, COOH, OH, alkoxy, NCOCH₃, or NR'R'', or two of the substituents R₁, R₂, R₃, or R₄ fom together a ring which is annealed to the benzthiazole moitie

R₅, R₆, R₇, R₈, independently from each other signifies H, Methoxy, Alkyl, Aryl, Alkoxy, SO₃H, SO₃R, Cl. Br. F, NR'R", SO₂-CH₂-CH₂-OH, or or two of the substituents R₅, R₆, R₇, or R₈, fom together a ring which is annealed to the phenyl group of the aminopyrazolo moitie.

R, R' and R'' are independently from each other H, alkyl or alkoxy or azylgroups groups.

and the salts and mixtuers thereof.

- 2. A process of dyeing or printing fibrous materials consisting of natural and/or synthetic polyamides by applying thereto dyestuffs according to formula I as defined in claim I, their salts or mixtures thereof.
- 3. Fibrous materials consisting of natural and/or synthetic polyamides dyed with dyestuffs of formula (I) as defined in claim 1, their salts or mixtures thereof.



The invention relates to novel azo dyestuffs, the use of such compounds and/or-mixturesfor printing recording materials, especially paper or papery substrates, textile fibre materials, plastic films and plastic transparencies by the inkjet printing process and also to the recording materials printed thereby.

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